Advanced Exergoenvironmental Analysis of a Near-Zero Emission Power Plant with Chemical Looping Combustion

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ABSTRACT: Carbon capture and storage (CCS) from power plants can be used to mitigate CO_2 emissions from the combustion of fossil fuels. However, CCS technologies are energy intensive, decreasing the operating efficiency of a plant and increasing its costs. Recently developed advanced exergy-based analyses can uncover the potential for improvement of complex energy conversion systems, as well as qualify and quantify plant component interactions. In this paper, an advanced exergoenvironmental analysis is used for the first time as means to evaluate an oxy-fuel power plant with CO_2 capture. The environmental impacts of each component are split into avoidable/unavoidable and endogenous/exogenous parts. In an effort to minimize the environmental impact of the plant operation, we focus on the avoidable part of the impact (which is also split into endogenous and exogenous parts) and we seek ways to decrease it. The results of the advanced exergoenvironmental analysis show that the majority of the environmental



impact related to the exergy destruction of individual components is unavoidable and endogenous. Thus, the improvement potential is rather limited, and the interactions of the components are of lower importance. The environmental impact of construction of the components is found to be significantly lower than that associated with their operation; therefore, our suggestions for improvement focus on measures concerning the reduction of exergy destruction and pollutant formation.

INTRODUCTION

With conventional exergy-based analyses, the locations, magnitudes, and causes of thermodynamic inefficiencies, costs, and environmental impacts are identified, and strategies for improvement can be found.¹⁻⁶ However, when component interactions are not considered, optimization strategies can be misguided, especially when complex systems with a large number of mutually affected components are considered. Advanced exergy-based analyses attempt to address this shortcoming.^{3,7–11}

Advanced exergy-based analyses have been developed at the Institute for Energy Engineering of the Technische Universität Berlin. In past publications, the exergoenvironmental analysis has only been partly presented, while advanced exergetic and exergoeconomic analyses have considered only relatively simple plants.^{8–11} This paper presents the results of the first application of an advanced exergoenvironmental analysis applied to a power plant incorporating CO₂ capture.

The calculations conducted in an advanced exergoenvironmental analysis depend on the results of an advanced exergetic analysis³ and are, to a large extent, analogous, as long as no significant component-related (i.e., construction-related) environmental impacts are found. In an advanced exergoenvironmental analysis, the environmental impacts associated with exergy destruction and pollutant formation are separated into avoidable/unavoidable, endogenous/exogenous, and the respective combined parts (e.g., avoidable endogenous, avoidable exogenous, etc.). In most cases, the component-related environmental impact is of low importance when compared to that of the operation of the plant (impact of exergy destruction) and it is therefore not analyzed in detail.

The plant considered here (Figure 1) is an oxy-fuel combined cycle power plant incorporating chemical looping combustion (CLC).^{12–17} This plant has already been examined using conventional exergy-based analyses.⁶ The structure and operating conditions of the plant are based on a reference plant that does not consider CO₂ capture.⁶ Here, the conventional exergoenvironmental analysis is reassessed, while the results of the advanced exergetic and exergoeconomic analyses for the plant can be found in ref 18.

METHODOLOGY: ADVANCED EXERGOENVIRONMENTAL ANALYSIS

In an advanced exergoenvironmental analysis, the environmental impacts associated with exergy destruction and pollutant formation (PF) are split based on their source.³ By considering different operating conditions, we can distinguish whether the environmental impacts of exergy destruction and/or pollutant formation are caused by the component itself (endogenous, $\dot{B}_{\mathrm{D},k}^{\mathrm{EN}}$, $\dot{B}_{k}^{\mathrm{PF,EN}}$) or by the operation of the remaining components (exogenous, $\dot{B}_{\mathrm{D},k}^{\mathrm{EX}}$, $\dot{B}_{k}^{\mathrm{PF,EX}}$). In more detail, the endogenous environmental impact of a component is associated

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Figure 1. Structure of the plant with chemical looping combustion, CLC (η : energetic efficiency, ϵ : exergetic efficiency).

with its operating conditions and the irreversibilities caused by its performance. Exogenous environmental impact, on the other hand, is the part of the impact of a considered component that exists due to the assigned role of the remaining plant components of the structure. Depending on the operation of the remaining components, the environmental impact of the component of interest will either increase or decrease.

The calculation of the endogenous environmental impacts is based on calculations conducted in an advanced exergetic analysis.^{3,7} To calculate the endogenous exergy destruction of component k, the component must operate under real conditions, while all other components of the process must operate without irreversibilities (i.e., theoretically). In all cases, the net power output of the overall plant is kept constant and equal to that of the initial simulation (real case).

$$\dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{EN}} = b_{\mathrm{F},\mathrm{k}}^{\mathrm{real}} \dot{E}_{\mathrm{D},\mathrm{k}}^{\mathrm{EN}} \tag{1}$$

$$\dot{B}_{k}^{\text{PF,EN}} = \sum_{i} b_{i}^{\text{PF}} (\dot{m}_{i,\text{out}} - \dot{m}_{i,\text{in}})^{\text{EN}}$$
(2)

 $b_{\rm F,k}^{\rm real}$ is the specific environmental impact of fuel of component k in the real case and $\dot{E}_{\rm D,k}^{\rm EN}$ is its endogenous exergy destruction. The difference $\dot{m}_{\rm i,out} - \dot{m}_{\rm i,in}$ in the calculation of the environmental impact of pollutant formation refers to the net mass flow of each pollutant i exiting the reactor (and emitted to the environment), and $b_i^{\rm PF}$ is its specific environmental impact. The mass flows of the pollutants in the endogenous operation of the reactor have been calculated in the simulations used for estimating the endogenous exergy destruction. Although the composition of the flue gases of the reactor remains the same, the different operating conditions result in changes in the mass flows in the overall plant. Thus, the absolute mass flows of the pollutants differ from those of

the real case. Here, the life cycle impact assessment method Eco-indicator 99 has been used, and the considered values of b_i^{PF} are shown in Table 1.¹⁹

Table 1. Eco-indicator Points

product system boundaries	pollutant quantity (kg/s)	eco-indicator (mPts/kg) ^a
natural gas (production and distribution)	0.28	180.0
CO ₂ emission	0.35	5.4
CH ₄ emission	0.28	114.6
NO _X emission	-	2749.4

^{*a*}1 Point (Pt) is representative of one thousandth of the yearly environmental load of one average European inhabitant (this value is calculated by dividing the total environmental load in Europe by the number of inhabitants and multiplying it by 1000).¹⁹

The exogenous parts of the environmental impacts $(\dot{B}_{D,k}^{EX}, \dot{B}_{k}^{PF,EX})$ are calculated by subtracting the endogenous from the real parts, $\dot{B}_{D,k}^{real}$ and $\dot{B}_{k}^{PF,real}$:

$$\dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{EX}} = \dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{real}} - \dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{EN}}$$
(3)

$$\dot{B}_{k}^{PF,EX} = \dot{B}_{k}^{PF,real} - \dot{B}_{k}^{PF,EN}$$
(4)

The exogenous impacts are therefore the impacts imposed on component k through the operation of the remaining n - 1 components that constitute the plant. The $\dot{B}_{D,k}^{EX}$ of component k can also be traced to the specific components that cause it, and it is calculated by examining the plant components in pairs (r+k). However, the sum of these individual exogenous environmental impacts differs from the exogenous impact of component k (calculated with eq 3). This leads to the definition of the

mexogenous environmental impact (MX, $\dot{B}_{D,k}^{MX}$), which represents the simultaneous interactions among the pairs of components examined and the remaining components of the process:⁷

$$\dot{B}_{D,k}^{MX} = \dot{B}_{D,k}^{EX} - \sum_{\substack{r=1\\r \neq k}}^{n} \dot{B}_{D,k}^{EX,r}$$
(5)

with

$$\sum_{\substack{r=1\\r\neq k}}^{n} \dot{B}_{D,k}^{\text{EX,r}} = \sum_{\substack{r=1\\r\neq k}}^{n} (\dot{B}_{D,k}^{\text{EN,r+k}} - \dot{B}_{D,k}^{\text{EN}})$$

The unavoidable environmental impact of exergy destruction of a component is calculated based on its unavoidable exergy destruction, $\dot{E}_{D,k}^{UN}$, and its specific environmental impact of fuel, $b_{F,k}^{real}$. The assumptions used to calculate the unavoidable exergy destruction include high efficiencies with minimum irreversibilities and are to some extent arbitrary, i.e., they depend on the decision maker.³ The unavoidable environmental impact of pollutant formation includes the CO₂ emissions, because complete combustion is assumed (in eq 7, i: CO₂).

$$\dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{UN}} = b_{\mathrm{F},\mathrm{k}}^{\mathrm{real}} \dot{E}_{\mathrm{D},\mathrm{k}}^{\mathrm{UN}} \tag{6}$$

$$\dot{B}_{k}^{\text{PF,UN}} = \sum_{i} b_{i}^{\text{PF}} (\dot{m}_{i,\text{out}} - \dot{m}_{i,\text{in}})$$
(7)

The avoidable impact of exergy destruction is obtained with eq 8, while the avoidable pollutant formation includes the NO_X and CH_4 emissions that could eventually be avoided by changing the combustion conditions:

$$\dot{B}_{D,k}^{AV} = \dot{B}_{D,k}^{real} - \dot{B}_{D,k}^{UN}$$
(8)

$$\dot{B}_{k}^{\text{PF,AV}} = b_{\text{NO}_{X}}^{\text{real}} \dot{m}_{\text{NO}_{X},\text{out}} + b_{\text{CH}_{4}}^{\text{real}} \dot{m}_{\text{CH}_{4},\text{out}}$$
(9)

The avoidable and unavoidable environmental impacts of component k can also be split into endogenous and exogenous parts as:

$$\dot{B}_{\mathrm{D},k}^{\mathrm{UN,EN}} = b_{\mathrm{F},k}^{\mathrm{real}} \dot{E}_{\mathrm{D},k}^{\mathrm{UN,EN}}$$
(10)

$$\dot{B}_{k}^{PF,UN,EN} = \dot{E}_{P,k}^{EN} \left(\frac{\dot{B}_{k}^{PF^{*}}}{\dot{E}_{P,k}} \right)^{UN}$$
(11)

$$\dot{B}_{\mathrm{D},k}^{\mathrm{UN,EX}} = \dot{B}_{\mathrm{D},k}^{\mathrm{UN}} - \dot{B}_{\mathrm{D},k}^{\mathrm{UN,EN}}$$
(12)

$$\dot{B}_{k}^{PF,UN,EX} = \dot{B}_{k}^{PF,UN} - \dot{B}_{k}^{PF,UN,EN}$$
(13)

The avoidable endogenous and exogenous impacts are then calculated by subtracting the corresponding unavoidable part from the total endogenous and exogenous environmental impacts, respectively:

$$\dot{B}_{D,k}^{AV,EN} = \dot{B}_{D,k}^{EN} - \dot{B}_{D,k}^{UN,EN}$$
 (14)

$$\dot{B}_{k}^{PF,AV,EN} = \dot{B}_{k}^{PF,EN} - \dot{B}_{k}^{PF,UN,EN}$$
(15)

$$\dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{AV,EX}} = \dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{EX}} - \dot{B}_{\mathrm{D},\mathrm{k}}^{\mathrm{UN,EX}}$$
(16)

$$\dot{B}_{k}^{PF,AV,EX} = \dot{B}_{k}^{PF,EX} - \dot{B}_{k}^{PF,UN,EX}$$
(17)

By now we can characterize specific parts of the environmental impacts of each plant component depending on their source and potential for mitigation. However, to identify the real improvement potential of individual plant components, the total avoidable environmental impact including both exergy destruction and pollutant formation must be calculated at the component level:

$$\dot{B}_{k}^{PF,AV,\sum} = \dot{B}_{k}^{PF,AV,EN} + \sum_{\substack{r=1\\r\neq k}}^{n} \dot{B}_{r}^{PF,AV,EX,k}$$
(18)

$$\dot{B}_{D,k}^{AV,\Sigma} = \dot{B}_{D,k}^{AV,EN} + \sum_{\substack{r=1\\r\neq k}}^{n} \dot{B}_{D,r}^{AV,EX,k}$$
(19)

Here,
$$\sum_{r=1}^{n} \dot{B}_{r}^{PF,AV,EX,k}$$
 and $\sum_{r=1}^{n} \dot{B}_{D,r}^{AV,EX,k}$ are the total avoidable
 $r \neq k$ $r \neq k$

environmental impacts of pollutant formation and exergy destruction of component r, respectively, caused by component k. The avoidable exogenous impact of exergy destruction is calculated as:

$$\dot{B}_{\mathrm{D,r}}^{\mathrm{AV,EX,k}} = b_{\mathrm{F,k}}^{\mathrm{real}} \dot{E}_{\mathrm{D,r}}^{\mathrm{AV,EX,k}}$$
(20)

where, $\dot{E}_{D,r}^{AV,EX,k}$ has been calculated for all components in a preceding advanced exergetic analysis.¹⁸

The avoidable exogenous environmental impact of pollutant formation caused by component k to each component r of the remaining plant components is calculated after the unavoidable exogenous environmental impact of pollutant formation, $B_r^{\text{PF,UN,EX,k}}$:

$$\dot{B}_{r}^{PF,AV,EX,k} = \dot{B}_{r}^{PF,EX,k} - \dot{B}_{r}^{PF,UN,EX,k}$$
(21)

The unavoidable exogenous environmental impact of pollutant formation is calculated through the unavoidable endogenous environmental impact, $\dot{B}_{\rm r}^{\rm PF\,\,UN\,\,EX}$:

$$\dot{B}_{r}^{PF,UN,EX,k} = \dot{B}_{r}^{PF,UN,EN,r+k} - \dot{B}_{r}^{PF,UN,EN}$$
(22)

where,
$$\dot{B}_{r}^{PF,UN,EN,r+k} = \dot{E}_{P,r}^{EN,r+k} \left(\frac{\dot{B}_{r}^{PF^{*}}}{\dot{E}_{P,r}}\right)^{ON}$$
, with $\dot{E}_{P,r}^{EN,r+k}$

equivalent to the $E_{P,r}$, when components r and k operate under real conditions and all remaining components operate under theoretical conditions.

The total avoidable environmental impacts (eqs 18 and 19) reveal the components with the largest influence on the overall plant. Taking action to improve the operation of these components should lead to a reduction in the environmental impact of the overall plant.

■ THE POWER PLANT

The oxy-fuel power plant considered in this paper includes chemical looping combustion²⁰⁻²⁵ for easy CO₂ capture (CLC plant). In this plant, the conventional combustion chamber is replaced by two reactors, an oxidizing or air reactor (AR), and a fuel reactor (FR). A metal oxide is used as a solid oxygen carrier (OC) between these two reactors; thus, no direct contact between the air and the fuel takes place. The two reactors are simulated here as a black box (CLC reactor). We assume that

	$\dot{B}_{{ m D},{ m k}}^{ m real}$	$\dot{B}_{{ m D},{ m k}}^{ m UN}$	$\dot{B}^{ m AV}_{ m D,k}$	$\dot{B}^{\mathrm{EN}}_{\mathrm{D,k}}$	$\dot{B}_{\mathrm{D,k}}^{\mathrm{EX}}$	$\dot{B}_{\mathrm{D,k}}^{\mathrm{AV,EN}}$	$\dot{B}_{\mathrm{D,k}}^{\mathrm{AV,EX}}$	$\dot{B}_{{ m D},{ m k}}^{{ m UN},{ m EN}}$	$\dot{B}_{\mathrm{D,k}}^{\mathrm{UN,EX}}$
C1	79.6	43.8	35.8	46.9	32.7	21.3	14.5	25.7	18.2
CLC	671.8	447.7	224.0	576.5	95.3	197.5	26.6	379.0	68.7
GT1	114.1	48.2	65.8	75.5	38.6	38.7	27.1	36.8	11.5
GT2	11.4	3.3	8.1	6.0	5.3	3.6	4.5	2.4	0.9
NGPH	21.1	0.5	20.5	12.1	9.0	0.6	20.0	11.5	-11.0
HPSH	11.6	7.6	4.0	3.8	7.7	0.5	3.5	3.3	4.3
HPEV	16.1	11.5	4.6	7.3	8.8	1.3	3.2	6.0	5.5
HPEC	21.3	15.8	5.5	10.8	10.5	1.8	3.7	9.0	6.8
RH	13.0	3.5	9.5	7.8	5.2	2.0	7.5	5.8	-2.4
IPEV	5.6	3.3	2.3	4.9	0.7	2.3	0.1	2.7	0.6
IPEC	1.9	1.3	0.6	1.2	0.7	-0.1	0.7	1.3	0.0
LPSH	1.8	0.6	1.2	1.0	0.8	0.7	0.5	0.3	0.4
LPEV	17.6	12.8	4.8	8.8	8.8	1.6	3.2	7.2	5.6
LPEC	21.8	12.2	9.6	13.1	8.7	6.6	3.0	6.5	5.7
HPST	11.9	4.6	7.3	5.6	6.3	3.0	4.3	2.6	2.0
IPST	9.8	4.3	5.4	5.6	4.2	2.5	3.0	3.1	1.2
LPST	48.0	25.4	22.6	31.3	16.6	13.0	9.6	18.3	7.0
ST4	35.0	5.1	29.8	19.4	15.5	16.6	13.2	2.8	2.3
C2	6.1	1.5	4.6	4.3	1.8	3.3	1.3	1.0	0.5
C3	6.4	1.5	4.9	4.7	1.8	3.6	1.3	1.1	0.4
C4	6.4	1.5	5.0	4.7	1.7	3.6	1.3	1.0	0.4
C5	6.7	1.5	5.2	4.9	1.8	3.8	1.4	1.0	0.4
FG COND	91.1			60.4	30.6				
COOL1	6.2			2.5	3.7				
COOL2	7.8			3.4	4.4				
COOL3	7.6			3.3	4.3				
COOL4	8.1			3.4	4.7				
COND	82.5			40.2	42.3				

in the FR, 98% of the methane reacts with oxygen transferred from the AR. The remaining unreacted 2% of the fuel is not recycled back to the FR but represents a loss. The air ratio, the ratio between the oxygen included in the air and the oxygen needed for stoichiometric combustion, is set to 2.9, which is required to achieve outlet temperatures of the air and the fuel reactors of 1200 °C and 930 °C, respectively. These temperatures are also the inlet temperatures of the expanders of the plant. It has been suggested that the inlet temperature of the CO_2/H_2O expander (GT2) should be as low as 900 °C, to increase the conversion of the fuel in the FR and the energy available for the oxidation of the metal in the AR.^{22,23} With this lower temperature, a lower cost for the expander is also achieved. When compared with various metals that have been suggested as oxygen carriers, Ni and its corresponding oxides generally show higher oxidation and reduction rates, as well as greater durability after many repeated cycles.^{26–29} Thus, here, a Ni-based OC is considered, and the reacted CH₄ is fully converted to CO₂.

RESULTS AND DISCUSSION

The CLC plant has been examined with conventional and advanced exergy-based environmental analyses. The conventional exergoenvironmental analysis has been performed using Eco-indicator 99.^{2,4,19} The data considered here are different from those presented in ref 4, because pollutant formation impacts have now been considered.³ In addition, the environmental impacts associated with exergy destruction and pollutant formation have been split. The component-related environmental impacts have been found to have a relatively negligible

influence on the total impact, and thus they have not been split further. Selected results of the advanced exergoenvironmental analysis at the component level are presented in Tables 2-5.

For the majority of the components, most of the environmental impact has been found to be unavoidable. Exceptions are the expander of the main gas turbine system (GT1), the high- and intermediate-pressure steam turbines (HPST and IPST), the CO_2 compressors (C1-C4), the additional steam turbine that drives the CO₂ compressors (ST4), GT2, and the natural gas preheater (NGPH). The high avoidable environmental impact of exergy destruction of the NGPH is associated with simulation assumptions. In the initial simulation, the preheater is considered together with a control valve, the exergy destruction within which is associated with pressure reduction. This exergy destruction is assumed to be avoidable when the preheater is considered in isolation. Changes to components with high avoidable environmental impact of exergy destruction can affect the overall plant significantly. Nevertheless, interactions among components should be considered in parallel to achieve improvement of the overall process. As shown in Table 2, most of the $\dot{B}_{D,k}$ is found to be endogenous, suggesting that component interactions are of lower significance. Specifically, for the component with the highest impact of exergy destruction, the CLC reactor, the endogenous impact is six times higher than the exogenous impact. Thus, the internal operation of this component has priority when considering improvement measures.

With advanced exergy-based methods, we reveal both positive and negative effects among plant components. A positive exogenous value of component k means that if the overall

Γable 3. Selected Results from	Splitting the Exogenous	Environmental Impact of	f Exergy Destruction ((mPts/s)"
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component, k	$\dot{B}_{{ m D},{ m k}}^{{ m EX}}$	component, r	$\dot{B}_{\mathrm{D,k}}^{\mathrm{EX,r}}$	component, k	$\dot{B}_{\mathrm{D,k}}^{\mathrm{EX}}$	component, r	$\dot{B}_{\mathrm{D,k}}^{\mathrm{EX,r}}$
CLC	95.27	C1	10.85	ST4	16.06	C1	0.72
		GT1	28.62			CLC	2.79
		ST4	3.80			GT1	0.83
		SUM	81.22 (70.89)			SUM	11.23 (5.24)
		MX^b	14.05			MX^b	4.30
C1	32.65	CLC	22.69	GT1	38.62	C1	2.00
		GT1	2.11			CLC	7.89
		ST4	0.28			ST4	0.46
		SUM	27.98 (22.92)			SUM	15.37 (64.16)
		MX^b	4.67			MX^b	23.25

^{*a*}In parentheses the sum of exergy destruction caused by component k to the remaining components r is shown. ^{*b*}Mexogenous environmental impact, as calculated with eq 5.

Table 4. Avoidable Environmental Impact of Exergy Destruction (mPts/s)

component, k	$\sum_{\substack{r=1\\r\neq k}}^{n} \dot{B}_{\mathrm{D},r}^{\mathrm{AV,EX,k}}$	$\dot{B}_{D,k}^{ m AV,EN}$	$\dot{B}_{\mathrm{D,k}}^{\mathrm{AV,\Sigma}}$
CLC	20.48 (9.4%)	197.48 (90.6%)	217.96
GT1	16.93 (30.4%)	38.70 (69.6%)	55.63
C1	1.31 (5.8%)	21.27 (94.2%)	22.58

impact of the other components is decreased, the impact of component k will be decreased as well. On the other hand, a negative exogenous value shows that a decrease in the environmental impact of the remaining plant components will have the opposite effect on component k, i.e., it would result in an increase in its environmental impact. Additionally, negative values of avoidable endogenous environmental impacts (e.g., the IPEC in Table 3) stem from negative values of the avoidable endogenous exergy destruction.³ Negative avoidable endogenous values are found when the endogenous is smaller than the unavoidable endogenous part of exergy destruction, which happens when the unavoidable surpasses the endogenous exergy destruction per exergy of product. This effect can only be explained through the mexogenous impacts that are a result of simultaneous interactions among all plant components. In this special case, due to component interactions, the considered component works more efficiently when it operates in the plant and with the conditions required for the calculation of its endogenous exergy destruction than when it operates in isolation and under the conditions required for the calculation of the unavoidable exergy destruction. Nonetheless, negative avoidable endogenous values are rare and they are only obvious when the component considered has small exergy destruction, and thus plays a less significant role in the structure.

Results from splitting the exogenous environmental impacts of exergy destruction for selected components are shown in Table 3. The sum of the columns refers to the total exogenous environmental impact of exergy destruction caused within the considered component k by the remaining components of the plant, while the values in the parentheses show the impact caused by component k to the remaining components of the plant. We find that the effect of the CLC reactor on the other components of the plant is of similar magnitude to the effect the other components exert on the reactor. These calculations are determined by the intense interactions among the three components of the GT system. On the other hand, GT1 is not influenced so much by the other components, but it influences the operation of the other components greatly. This happens because the expander directly influences the operation of the majority of the components in the steam cycle, since its operation determines the thermal energy available through the flue gases. In summary, the highest exogenous impact is caused by the components of the gas turbine system, i.e., the reactors, followed by GT1 and C1. These components influence each other substantially: 41% of the impact of the reactors is caused by C1 and GT1, while a large part of the impact imposed on C1 and GT1 stems from the reactors.

The total avoidable environmental impact of exergy destruction has been calculated based on eq 19, and the results of the most influential components of the plant are shown in Table 4. The CLC reactor presents relatively high avoidable endogenous and exogenous values, when compared to the other components, resulting in a higher overall impact. GT1 is ranked second, C1 third, and ST4 fourth. A large difference between C1 and GT1 is the relatively larger avoidable exogenous value of GT1, because C1 influences the remaining components of the plant less. In general, it has been found that the components with the highest avoidable values are also those with the highest influence on the plant. Thus, improving these components will also improve the operation of other components and lead to a significant reduction in the environmental impact of the overall plant.

In summary, the components of the plant with the highest influence and avoidable impacts are the components constituting the GT system. The CLC reactor is the most important component with the highest environmental impact of exergy destruction, although it also has the highest unavoidable impact. This component has an approximately four times higher total avoidable impact than GT1. Since the total environmental impact of the CLC reactor is mainly determined by its exergy destruction, measures to improve the mixing of streams and to minimize temperature differences and pressure losses have to be considered. Additionally, the avoidable impact of pollutant formation of the CLC reactor is mostly endogenous and can,

The of opinically the birth on the of the official the official to the official the official to the official t	Table	5.	Splitting	the	Environmental	Impact	of]	Pollutant	Formation	(mPts/s	s)
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	$\dot{B}_{ m k}^{ m PF,real}$	$\dot{B}_{ m k}^{ m PF,~UN}$	$\dot{B}_{ m k}^{ m PF,AV}$	$\dot{B}_{ m k}^{ m PF,EN}$	$\dot{B}_{ m k}^{ m PF,EX}$	$\dot{B}_{ m k}^{ m PF,AV,EN}$	$\dot{B}_{ m k}^{ m PF,AV,EX}$	$\dot{B}_{ m k}^{ m PF,UN,EN}$	$\dot{B}_{ m k}^{ m PF,UN,EX}$
CLC	237.89	205.89	32.09	206.46	31.44	32.24	-0.14	174.22	31.58

therefore, be decreased through changes in its operating conditions. The results from splitting the environmental impact of pollutant formation within the reactors of the plant are shown in Table 5. Overall, the majority of the environmental impact related to the exergy destruction is unavoidable and endogenous. Thus, the interactions of the components are of secondary importance. Nevertheless, changes to both component interactions and/or avoidable/ unavoidable impacts can influence one another and should be considered in parallel when an overall improvement is desired.

 CO_2 capture in combined cycle power plants is generally a costly process and only small improvements of the presented CLC process are possible. However, the application of improvement steps suggested by the advanced exergy-based analyses should lead to an improved performance of the overall plant.

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NOMENCLATURE

b	exergy-specific (mPts/MJ) or mass-specific (mPts/kg)
	environmental impact
<i>B</i> −	environmental impact (mPts/s)
Ė	exergy rate (MW)
Super-/S	ubscripts
AV	avoidable
D	exergy destruction
EN	endogenous
EX	exogenous
F	fuel (exergy)
Р	product (exergy)
i,j	stream
k,r	component
L	loss
PF	pollutant formation
real	calculated when components operate under real
	conditions
UN	unavoidable
\sum	sum of the component-individual exogenous envi-
	ronmental impacts
Abbrevia	tions
AR	air reactor
C (1–5)	compressor
CCS	carbon capture and storage
CLC	chemical looping combustion
COND	condenser
COOL	cooler
СТ	cooling tower
DB	duct burner
EC	economizer
EV	evaporator
FG	flue gas
FR	fuel reactor

GT gas turbine

- HP high pressure
- IP intermediate pressure
- LP low pressure LCA life cycle assessment
- NG natural gas
- OC oxygen carrier
- PH preheater
- RH reheater
- SH superheater
- ST steam turbine

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